

REFERENCES

- Amjad, Z. Influence of Polyelectrolytes on the Precipitation of Amorphous Calcium Phosphate. Colloids and Surfaces 48 (1990) : 95 - 106.
- Chang, D., and Hwang, S. Removal of Metal Ions from Liquid Solutions by Crossflow Microfiltration. Separation Science and Technology 31 (1996): 1831- 1842.
- Chang, L.Y. An Industrial Wastewater Pollution Prevention Study Evaluation of Precipitation and Separation Process. Colloids and surfaces 53 (1991): 241 - 255.
- Cheryan, M. Ultrafiltration Handbook PA : Technomic Publishing, 1986.
- Christian, S.D., Tucker, E.E., and Scamehorn, J.F. Colloid - Enhanced Ultrafiltration Process for Purifying Aqueous Streams and Reservoirs. American Environment Laboratory 2(1990): 13.
- Porter, M.C. Handbook of Separation Techniques for Chemical Engineers New York: Mc. Graw - Hill, 1979.
- Sasaki, K.J., Burnett, S.L., Christian, S.D., and Tucker, E.E. Use of Micellar - Enhanced Ultrafiltration of Multivalent Ions. Removal of Cu²⁺ by Sodium Poly(styrene sulfonate). Langmuir 5 (1989): 363 - 369.
- Scamehorn, J.F., Christian, S.D., Tucker, E.E., and Tan, B.I. Concentration Polarization in Polyelectrolyte - Enhanced Ultrafiltration. Colloids and Surfaces 49 (1990) : 259 - 267.
- Sriratana, S., Scamehorn, J.F., Chavadej, S., Saiwan, C., Haller, K.J., Christian, S.D., and Tucker, E.E. Use of polyelectrolyte - enhanced ultrafiltration to remove chromate from water. Separation Science and

- Technology 31 (1996) : 2493 - 2504.
- Tabatabai, A., Scamehorn, J.F., and Christian, S.D. Economic feasibility study of polyelectrolyte - enhanced ultrafiltration (PEUF) for water softening. Journal of Membrane Science 100 (1995) : 193 - 207.
- Tabatabai, A., Scamehorn, J.F., and Christian, S.D. Water Softening Using Polyelectrolyte - Enhanced Ultrafiltration. Separation Science and Technology 30 (1995) : 211 - 224.
- Tucker, E.E., Christian, S.D., Scamehorn, J.F., Chiyama, H.U., and Guo, W. Transport and Remediation of Subsurface Contaminants. Washington D.C.: American Chemical Society, 1992.
- Tuncay, M., Christian, S.D., Tucker, E.E., Taylor, R.W., and Scamehorn, J.F. Ligand - Modified Polyelectrolyte - Enhanced Ultrafiltration with Electrostaic Attachment of Ligands. 1. Removal of Cu (II) and Pb (II) with Expulsion of Ca (II). Langmuir 10 (1994) : 4688 - 4692.
- Tuncay, M., Christian, S.D., Tucker, E.E., Taylor, R.W., and Scamehorn, J.F. Ligand - Modified Polyelectrolyte - Enhanced Ultrafiltration with Electrostaic Attachment of Ligands. 2. Use of Diethylenetriaminepentaacetic Acid/Cationic Polyelectrolyte Mixtures to Remove both Cations and Anions from Aqueous Streams. Langmuir 10 (1994): 4693 - 4697.
- Uchiyama, H., Christian, S.D., and Tucker, E.E. Solubilization and Separation of p- tert - butylphenol Using Polyelectrolyte / Surfactant complexes in Colloid - Enhanced Ultrafiltration. Journal of Colloid and interface science 163 (1994): 493.
- Wesselingh, J.A., and Vonk, P. Ultrafiltration of a large polyelectrolyte. Journal of Membrane Science 99 (1995): 21-27.

APPENDIX A

Experimental conditions

APPENDIX A.1 Study of the effect of increasing chromate concentration on the barium chromate precipitation (Section 4.2)

Type no.	[QUAT] (M)	[CrO ₄ ²⁻] (M)	[Ba ²⁺] (M)	[CrO ₄ ²⁻] [QUAT]	[Ba ²⁺] [CrO ₄ ²⁻]
a-1-1	0.1	0.08	0.08	0.8	1
a-1-2	0.1	0.04	0.08	0.4	2
a-1-3	0.1	0.016	0.08	0.16	5
a-1-4	0.1	0.01	0.08	0.1	8
a-1-5	0.1	0.008	0.08	0.08	10
a-2-1	0.2	0.08	0.08	0.4	1
a-2-2	0.2	0.04	0.08	0.2	2
a-2-3	0.2	0.016	0.08	0.08	5
a-2-4	0.2	0.01	0.08	0.05	8
a-2-5	0.2	0.008	0.08	0.04	10
a-3-1	0.3	0.08	0.08	0.2667	1
a-3-2	0.3	0.04	0.08	0.1333	2
a-3-3	0.3	0.016	0.08	0.0533	5
a-3-4	0.3	0.01	0.08	0.0333	8
a-3-5	0.3	0.008	0.08	0.0267	10
a-4-1	0.4	0.08	0.08	0.2	1
a-4-2	0.4	0.04	0.08	0.1	2
a-4-3	0.4	0.016	0.08	0.04	5

Type no.	[QUAT] (M)	$[\text{CrO}_4^{2-}]$ (M)	$[\text{Ba}^{2+}]$ (M)	$[\text{CrO}_4^{2-}]$ [QUAT]	$[\text{Ba}^{2+}]$ $[\text{CrO}_4^{2-}]$
a-4-4	0.4	0.01	0.08	0.025	8
a-4-5	0.4	0.008	0.08	0.02	10
a-5-1	0	0.08	0.088	0	1
a-5-2	0	0.04	0.08	0	2
a-5-3	0	0.016	0.08	0	5
a-5-4	0	0.01	0.08	0	8
a-5-5	0	0.008	0.08	0	10

Note: The barium concentration was held constant at 0.08 M while the QUAT concentrations were constant at 0.1, 0.2, 0.3, and 0.4 M. The experimental data were shown in appendix B.2.

APPENDIX A.2 Study of the effect of increasing barium concentration on
the barium chromate precipitation (Section4.2)

Type no.	[QUAT] (M)	[CrO ₄ ²⁻] (M)	[Ba ²⁺] (M)	[CrO ₄ ²⁻] [QUAT]	[Ba ²⁺] [CrO ₄ ²⁻]
b-1-1	0.1	0.002	0.002	0.02	1
b-1-2	0.1	0.002	0.004	0.02	2
b-1-3	0.1	0.002	0.01	0.02	5
b-1-4	0.1	0.002	0.016	0.02	8
b-1-5	0.1	0.002	0.02	0.02	10
b-2-1	0.1	0.02	0.02	0.2	1
b-2-2	0.1	0.02	0.04	0.2	2
b-2-3	0.1	0.02	0.1	0.2	5
b-2-4	0.1	0.02	0.16	0.2	8
b-2-5	0.1	0.02	0.20	0.2	10
b-3-1	0.1	0.06	0.06	0.6	1
b-3-2	0.1	0.06	0.12	0.6	2
b-3-3	0.1	0.06	0.3	0.6	5
b-3-4	0.1	0.06	0.48	0.6	8
b-3-5	0.1	0.06	0.6	0.6	10
b-4-1	0	0.002	0.002	0	1
b-4-2	0	0.002	0.004	0	2
b-4-3	0	0.002	0.01	0	5
b-4-4	0	0.002	0.016	0	8
b-4-5	0	0.002	0.02	0	10
b-5-1	0	0.02	0.02	0	1
b-5-2	0	0.02	0.04	0	2

Type no.	[QUAT] (M)	$[\text{CrO}_4^{2-}]$ (M)	$[\text{Ba}^{2+}]$ (M)	$[\text{CrO}_4^{2-}]$ [QUAT]	$[\text{Ba}^{2+}]$ $[\text{CrO}_4^{2-}]$
b-5-3	0	0.02	0.1	0	5
b-5-4	0	0.02	0.16	0	8
b-5-5	0	0.02	0.2	0	10
b-6-1	0	0.06	0.06	0	1
b-6-2	0	0.06	0.12	0	2
b-6-3	0	0.06	0.3	0	5
b-6-4	0	0.06	0.48	0	8
b-6-5	0	0.06	0.6	0	10

Note: The QUAT concentration was held constant at 0.1 M while the chromate concentrations were constant at 0.002, 0.02, and 0.06 M. The experimental data were shown in appendix B.3.

APPENDIX A.3 Study of the effect of increasing QUAT concentration on
the barium chromate precipitation (Section4.2)

Type no.	[QUAT] (M)	[CrO ₄ ²⁻] (M)	[Ba ²⁺] (M)	[CrO ₄ ²⁻] [QUAT]	[Ba ²⁺] [CrO ₄ ²⁻]
c-1-1	0.1	0.02	0.02	0.2	1
c-1-2	0.2	0.02	0.02	0.1	1
c-1-3	0.3	0.02	0.02	0.0667	1
c-1-4	0.4	0.02	0.02	0.05	1
c-2-1	0.1	0.02	0.04	0.2	2
c-2-2	0.2	0.02	0.04	0.1	2
c-2-3	0.3	0.02	0.04	0.0667	2
c-2-4	0.4	0.02	0.04	0.05	2
c-3-1	0.1	0.02	0.1	0.2	5
c-3-2	0.2	0.02	0.1	0.1	5
c-3-3	0.3	0.02	0.1	0.0667	5
c-3-4	0.4	0.02	0.1	0.05	5
c-4-1	0.1	0.02	0.16	0.2	8
c-4-2	0.2	0.02	0.16	0.1	8
c-4-3	0.3	0.02	0.16	0.0667	8
c-4-4	0.4	0.02	0.16	0.05	8
c-5-1	0.1	0.02	0.2	0	10
c-5-2	0.2	0.02	0.2	0	10
c-5-3	0.3	0.02	0.2	0	10
c-5-4	0.4	0.02	0.2	0	10
c-6-1	0	0.02	0.02	0	1
c-6-2	0	0.02	0.04	0	2

Type no.	[QUAT] (M)	[CrO ₄ ²⁻] (M)	[Ba ²⁺] (M)	[CrO ₄ ²⁻] [QUAT]	[Ba ²⁺] [CrO ₄ ²⁻]
c-6-3	0	0.02	0.1	0	5
c-6-4	0	0.02	0.16	0	8
c-6-5	0	0.02	0.2	0	10

Note: The chromate concentration was held constant at 0.02 M while the barium concentrations were constant at 0.02, 0.04, 0.1, 0.16, and 0.2 M. The experimental data were shown in appendix B.4.

APPENDIX B

Experimental data

APPENDIX B.1 The experimental data of the equilibrium time study

Time(hrs)	CrO_4^{2-} Precipitates(%)	
	At [QUAT] = 0.4M [CrO ₄ ²⁻] = 0.8M [Ba ²⁺] = 0.4 M	At [QUAT] = 0.1M [CrO ₄ ²⁻] = 0.002M [Ba ²⁺] = 0.002M
0	48.4	89
0.5	57.72	92.05
1	45.84	91.3
2	51.78	91.21
4	46.67	93.78
6	52.21	88.02
10	53.41	89.99
12	52.25	86.44
15	52.18	85.62
24	46.58	86.95
48	46.47	85.97
96	45.67	86.74
144	45.05	85.63
168	46.97	

Note: The plots of the percentage of chromate in the barium chromate precipitate were shown in Figure 4-1.

APPENDIX B.2 Analytical data of the experiment obtained from appendix A.1

Type no.	Initial condition				[Ba ²⁺]				[CrO ₄ ²⁻]			
	[QUAT] (M)	[CrO ₄ ²⁻] / [QUAT]	[CrO ₄ ²⁻] (M)	[Ba ²⁺]/[CrO ₄ ²⁻]	Initial (ppm)	Supernatant (ppm)	Precipitate (ppm)	% Precipitate	Initial (ppm)	Supernatant (ppm)	Precipitate (ppm)	% Precipitate
a-1-1	0.1	0.800	0.080	1	10987.2	0.6803	10986.5197	99.99	9279.488	416.0690	8863.4190	95.52
a-1-2	0.1	0.400	0.040	2	10987.2	5797.0588	5190.1412	47.24	4639.744	0.4657	4639.2783	99.99
a-1-3	0.1	0.160	0.016	5	10987.2	9197.0588	1790.1412	16.29	1855.8976	0.2329	1855.6647	99.99
a-1-4	0.1	0.100	0.010	8	10987.2	9844.1177	1143.0823	10.40	1159.936	0.0000	1159.9360	100.00
a-1-5	0.1	0.080	0.008	10	10987.2	11012.5000	-25.3000	-0.23	927.9488	0.0000	927.9488	100.00
a-2-1	0.2	0.400	0.080	1	10987.2	2.4615	10984.7385	99.98	9279.488	373.7743	8905.7137	95.97
a-2-2	0.2	0.200	0.040	2	10987.2	5993.7107	4993.4893	45.45	4639.744	1.0081	4638.7359	99.98
a-2-3	0.2	0.080	0.016	5	10987.2	9773.5849	1213.6151	11.05	1855.8976	0.4911	1855.4065	99.97
a-2-4	0.2	0.050	0.010	8	10987.2	11303.1250	-315.9250	-2.88	1159.936	0.0000	1159.9360	100.00
a-2-5	0.2	0.040	0.008	10	10987.2	11135.2201	-148.0201	-1.35	927.9488	0.0000	927.9488	100.00
a-3-1	0.3	0.267	0.080	1	10987.2	1.8447	10985.3553	99.98	9279.488	467.7073	8811.7807	94.96
a-3-2	0.3	0.133	0.040	2	10987.2	5202.3121	5784.8879	52.65	4639.744	0.5488	4639.1952	99.99
a-3-3	0.3	0.053	0.016	5	10987.2	8537.5723	2449.6277	22.30	1855.8976	0.2744	1855.6232	99.99
a-3-4	0.3	0.033	0.010	8	10987.2	10251.4451	735.7549	6.70	1159.936	0.2744	1159.6616	99.98
a-3-5	0.3	0.027	0.008	10	10987.2	10557.8035	429.3965	3.91	927.9488	0.2744	927.6744	99.97
a-4-1	0.4	0.200	0.080	1	10987.2	3.4568	10983.7432	99.97	9279.488	597.2231	8682.2649	93.56
a-4-2	0.4	0.100	0.040	2	10987.2	5470.5882	5516.6118	50.21	4639.744	0.6401	4639.1039	99.99
a-4-3	0.4	0.040	0.016	5	10987.2	9147.0588	1840.1412	16.75	1855.8976	0.6401	1855.2575	99.97
a-4-4	0.4	0.025	0.010	8	10987.2	9158.8235	1828.3765	16.64	1159.936	0.6401	1159.2959	99.94
a-4-5	0.4	0.020	0.008	10	10987.2	9605.8824	1381.3176	12.57	927.9488	1.2802	926.6686	99.86
a-5-1	0	0.000	0.080	1	10987.2	1.2270	10985.9730	99.99	9279.488	139.2798	9140.2082	98.50
a-5-2	0	0.000	0.040	2	10987.2	5788.3436	5198.8564	47.32	4639.744	0.1902	4639.5538	100.00
a-5-3	0	0.000	0.016	5	10987.2	9898.7730	1088.4270	9.91	1855.8976	0.1432	1855.7544	99.99
a-5-4	0	0.000	0.010	8	10987.2	10576.6871	410.5129	3.74	1159.936	0.1902	1159.7458	99.98
a-5-5	0	0.000	0.008	10	10987.2	10773.0061	214.1939	1.95	927.9488	0.1667	927.7821	99.98

APPENDIX B.3 Analytical data of the experiment obtained from appendix A.2

Type no.	Initial condition				[Ba ²⁺]				[CrO ₄ ²⁻]			
	[CrO ₄ ²⁻] / [QUAT]	[CrO ₄ ²⁻]	[Ba ²⁺] / [CrO ₄ ²⁻]	[Ba ²⁺]	Initial (ppm)	Supernatant (ppm)	Precipitate (ppm)	Precipitate (%)	Initial (ppm)	Supernatant (ppm)	Precipitate (ppm)	Precipitate (%)
b-1-1	0.02	0.002	1	0.002	274.6800	2.0000	272.6800	99.27	231.9872	73.9995	157.9877	68.10
b-1-2	0.02	0.002	2	0.004	549.3600	171.3333	378.0267	68.81	231.9872	3.2494	228.7378	98.60
b-1-3	0.02	0.002	5	0.010	1373.4000	1096.3855	277.0145	20.17	231.9872	0.9218	231.0654	99.60
b-1-4	0.02	0.002	8	0.016	2197.4400	1971.0843	226.3557	10.30	231.9872	1.4058	230.5814	99.39
b-1-5	0.02	0.002	10	0.020	2746.8000	2580.7229	166.0771	6.05	231.9872	0.6914	231.2958	99.70
b-2-1	0.20	0.020	1	0.020	2746.8000	8.2667	2738.5333	99.70	2319.8720	42.3809	2277.4911	98.17
b-2-2	0.20	0.020	2	0.040	5493.6000	2962.6506	2530.9494	46.07	2319.8720	1.4519	2318.4201	99.94
b-2-3	0.20	0.020	5	0.100	13734.0000	11689.6552	2044.3448	14.89	2319.8720	0.2535	2319.6185	99.99
b-2-4	0.20	0.020	8	0.160	21974.4000	21086.2069	888.1931	4.04	2319.8720	0.4840	2319.3880	99.98
b-2-5	0.20	0.020	10	0.200	27468.0000	24500.0000	2968.0000	10.81	2319.8720	0.2535	2319.6185	99.99
b-3-1	0.60	0.060	1	0.060	8240.4000	13.3333	8227.0667	99.84	6959.6160	68.5147	6891.1013	99.02
b-3-2	0.60	0.060	2	0.120	16480.8000	8574.7126	7906.0874	47.97	6959.6160	0.9679	6958.6481	99.99
b-3-3	0.60	0.060	5	0.300	41202.0000	36408.0460	4793.9540	11.64	6959.6160	0.2535	6959.3625	100.00
b-3-4	0.60	0.060	8	0.480	65923.2000	64674.5562	1248.6438	1.89	6959.6160	0.4840	6959.1320	99.99
b-3-5	0.60	0.060	10	0.600	82404.0000	80414.2012	1989.7988	2.41	6959.6160	0.2535	6959.3625	100.00

APPENDIX B.4 Analytical data of the experiment obtained from appendix A.3

Type no.	Initial condition				[Ba ²⁺]				[CrO ₄ ²⁻]			
	[QUAT]	[CrO ₄ ²⁻] / [QUAT]	[Ba ²⁺] / [CrO ₄ ²⁻]	[Ba ²⁺]	Initial (ppm)	Supernatant (ppm)	Precipitate (ppm)	Precipitate (%)	Initial (ppm)	Supernatant (ppm)	Precipitate (ppm)	Precipitate (%)
c-1-1	0.1	0.20	1	0.02	2746.8	1.9231	2744.8769	99.93	2319.8720	151.8855	2167.9865	93.45
c-1-2	0.2	0.10	1	0.02	2746.8	1.5267	2745.2733	99.94	2319.8720	232.7032	2087.1688	89.97
c-1-3	0.3	0.07	1	0.02	2746.8	0.9091	2745.8909	99.97	2319.8720	322.1072	1997.7648	86.12
c-1-4	0.4	0.05	1	0.02	2746.8	1.2195	2745.5805	99.96	2319.8720	417.4501	1902.4219	82.01
c-2-1	0.1	0.20	2	0.04	5493.6	2711.9761	2781.6239	50.63	2319.8720	1.3358	2318.5362	99.94
c-2-2	0.2	0.10	2	0.04	5493.6	2527.9762	2965.6238	53.98	2319.8720	1.4596	2318.4124	99.94
c-2-3	0.3	0.07	2	0.04	5493.6	2346.4706	3147.1294	57.29	2319.8720	1.8458	2318.0262	99.92
c-2-4	0.4	0.05	2	0.04	5493.6	2308.8235	3184.7765	57.97	2319.8720	0.6079	2319.2641	99.97
c-3-1	0.1	0.20	5	0.10	13734.0	11131.2500	2602.7500	18.95	2319.8720	1.1132	2318.7588	99.95
c-3-2	0.2	0.10	5	0.10	13734.0	11309.3750	2424.6250	17.65	2319.8720	0.7298	2319.1422	99.97
c-3-3	0.3	0.07	5	0.10	13734.0	11434.3750	2299.6250	16.74	2319.8720	0.7911	2319.0809	99.97
c-3-4	0.4	0.05	5	0.10	13734.0	11700.0000	2034.0000	14.81	2319.8720	0.6079	2319.2641	99.97
c-4-1	0.1	0.20	8	0.16	21974.4	20977.0115	997.3885	4.54	2319.8720	0.4453	2319.4267	99.98
c-4-2	0.2	0.10	8	0.16	21974.4	19482.7586	2491.6414	11.34	2319.8720	0.2433	2319.6287	99.99
c-4-3	0.3	0.07	8	0.16	21974.4	21390.8046	583.5954	2.66	2319.8720	0.2637	2319.6083	99.99
c-4-4	0.4	0.05	8	0.16	21974.4	20620.6897	1353.7103	6.16	2319.8720	0.0000	2319.8720	100.00
c-5-1	0.1	0.20	10	0.20	27468.0	25208.3333	2259.6667	8.23	2319.8720	1.7811	2318.0909	99.92
c-5-2	0.2	0.10	10	0.20	27468.0	25440.4762	2027.5238	7.38	2319.8720	0.2433	2319.6287	99.99
c-5-3	0.3	0.07	10	0.20	27468.0	27821.4286	-353.4286	-1.29	2319.8720	0.5537	2319.3183	99.98
c-5-4	0.4	0.05	10	0.20	27468.0	27345.2381	122.7619	0.45	2319.8720	0.3039	2319.5681	99.99
c-6-1	0		1	0.20	27468.0	25.1515	27442.8485	99.91	2319.8720	5.6164	2314.2556	99.76
c-6-2	0		2	0.04	5493.6	2872.7273	2620.8727	47.71	2319.8720	1.6906	2318.1814	99.93
c-6-3	0		5	0.10	13734.0	12296.9697	1437.0303	10.46	2319.8720	0.2384	2319.6336	99.99
c-6-4	0		8	0.16	21974.4	22703.0303	-728.6303	-3.32	2319.8720	0.3959	2319.4761	99.98
c-6-5	0		10	0.20	27468.0	28878.7879	-1410.7879	-5.14	2319.8720	0.3718	2319.5002	99.98

CURRICULUM VITAE

Name : Sirirat Chaisin

Birth Date : September 20, 1968

Nationality : Thai

University Education :

1987 - 1990 Bachelor's Degree of Engineering in Chemical
Engineering, Prince of Songkla University.

Working Experience :

1990 - 1992 Maintenance Engineer, Yong Thai Chemical Co.,Ltd
1992 - 1995 Technical Engineer, National Thai Co.,Ltd