

REFERENCES

- Berkovitch-Yellin, Z., Van Mil, J., Addadi, L., Idelson, M., Lahav, M. and Leiserowitz, L. Crystal morphology engineering by "Tailor-Made" Inhibitors: A new probe to fine intermolecular interactions J. Am.Chem.Soc. 107 (1985): 3111.
- Browning, F.H. and Fogler, H.S. Precipitation and dissolution of calcium-phosphonates for the enhancement of squeeze lifetimes SPE 25164 presented at the Intl. Symposium on Oilfield chemistry, held in New Orleans, LA, 2-5 March, 1993, 137-147.
- Browning, F.H. and Fogler, H.S. Effect of synthesis parameters on the properties of calcium phosphonate precipitates Langmuir 11 (1995): 4143-4152.
- Browning, F.H. and Fogler, H.S. Effect of precipitation conditions on the formation of calcium-HEDP precipitates Langmuir 12 (1996): 5231-5238.
- Graham, G.M., Boak, L.S. and Sorbie, K.S. The influence of formation calcium on the effectiveness of generically different barium sulphate oilfield scale inhibitors SPE 37273 presented at the SPE Int. Symposium on Oilfield Chemistry, held in Houston, TX, 18-21 February, 1997, 611-626.
- Hong, S.A., Shuler, P.J., A mathematical model for the scale-inhibitor squeeze process SPE Production Engineering (November 1988): 597-607.

- Kan, a.T., Oddo, J.E. and Tomson, M.B. Formation of two calcium Diethylenetriaminepentakis(methylene phosphonic acid) precipitates and their physical chemical properties Langmuir (10), 1994: 1450-1455.
- Liu S.T. and Nancollas, G.H. Scanning electron microscope and kinetic studies of the crystallization and dissolution of barium sulfate crystals J. Crystal Growth 76 (1986): 57.
- Malandrino, A., Yuan, M.D., Sorbie, K.S. and Jordan, M.M. Mechanistic study and modeling of precipitation scale inhibitor squeeze processes SPE 29001 presented at the Intl. Symposium on Oilfield Chemistry, held in San Antonio, TX, 14-17 February, 1995, 597-612.
- Meyers, K.O., Skillman, H.L., Herring, G.D. Control of formation damage at Prudhoe bay, Alaska, by inhibitor squeeze treatment Journal of Petroleum Technology (June 1985): 1019-1039.
- Monsanto An introductory guide of Dequest Phosphonates by Monsanto Monsanto Technical Bulletin Publication No. 745 9151A.
- Monsanto Information on Dequest 2000 and 2006 Phosphonates Monsanto Technical Bulletin Publication No. 9023.
- Nancollas, H.G., Kazmierczak, T.F. and Schuttriger E. A controlled composition study of calcium carbonate crystal growth: The influence of scale inhibitors Corrosion-NACE (37) :76-81.
- Nancollas, G.H. Oilfield scale: physical chemical studies of its formation and prevention Chemicals in the Oil Industry (1985): 143-164.
- Oddo, J.E., and Tomson, M.B. The solubility and stoichiometry of calcium-DTPMP at 70 C in brine solutions at 4.7 and 5.0 Applied Geochemistry 5 (1990): 527-532.

- Oddo, J.E. and Tomson, M.B. Why scale forms and how to predict it SPE Production & Facilities (February 1994): 47-54.
- Rerkpattanapipat, P. Precipitation and dissolution of calcium-phosphonate on inhibition of scale formation in porous media. Master's Thesis, The Petroleum and Petrochemical College, Chulalongkorn University, 1996.
- Sorbie, K.S., Yuan, M.D., Chen, P., Todd, A.C. and Wat, R.M.S. The effect of pH on the adsorption and transport of phosphonate scale inhibitor through porous media SPE 25165 presented at the Intl. Symposium on Oilfield Chemistry, held in New Orleans, LA, 2-5 March, 1993, 149-166.
- Sorbie, K.S., Wat, R.M.S. and Todd, A.C. Interpretation and theoretical modeling of scale-inhibitor/tracer corefloods SPE Production Eng. 7 (1992): 307-312.
- Vetter, O.J. An evaluation of scale inhibitors Journal of Petroleum Technology 24 (August 1972): 997-1006.
- Yuan, M.D., Sorbie, K.S., Todd, A.C., Atkinson, L.M., Riley, H. and Gurden, S. The modelling of adsorption and precipitation scale inhibitor squeeze treatment in North Sea fields SPE 25163 presented at the Intl. Symposium on Oilfield Chemistry, held in New Orleans, LA, 2-5 March, 1993, 121-136.

APPENDIX

Experimental Data

A. Conditions for Batch Synthesis

Ca/ATMP Molar Ratio in Solution	Conc. of Calcium (M)	Amount of Calcium used (ml)	Conc. of ATMP (M)	Amount of ATMP used (ml)
1:1	0.5	57.7	0.6688	42.3
10:1	2.235	40	0.149	60

B. Conditions of precipitates in Micromodel Experiment

Type of precipitate	pH	Ca/phosphonate Molar Ratio in Solution	Conc. of Calcium (M)	Amount of Calcium used (ml)	Conc. of phosphonate (M)	Amount of phosphonate used (ml)
1:1 Ca-HEDP	1.5	1:1	0.1	72.9	0.2427	30
1:1 Ca-ATMP	1.5	1:1	0.5	33.44	0.1672	100
3:1 Ca-ATMP	9.0	1:1	0.5	57.7	0.6688	42.3

C. Elution curves in Micromodel Experiment

i) Elution curve of 3:1 Ca-ATMP Precipitate

Elution fluid : DI water, pH 9.0

Flowrate : 0.05 ml/min

Time (min)	PV	ATMP Conc. (ppm)
10.00	3.33	59311.93
20.00	6.67	4030.28
30.00	10.00	80.95
40.00	13.33	38.88
50.00	16.67	29.00
60.00	20.00	26.25
70.00	23.33	17.79
80.00	26.67	19.99
90.00	30.00	16.26
100.00	33.33	14.83
200.00	66.67	11.31
300.00	100.00	8.57
600.00	200.00	6.70
800.00	266.67	5.27
950.00	316.67	6.37
1200.00	400.00	6.92
1300.00	433.33	6.37
1400.00	466.67	4.91
1500.00	500.00	4.06
1700.00	566.67	13.73
1800.00	600.00	5.38
1900.00	633.33	6.59
2000.00	666.67	7.91
2050.00	683.33	7.44
2070.00	690.00	7.84

ii) Elution curves of 1:1 Ca-HEDP Precipitates

Elution fluids : DI water, 0.1 M CaCl₂, 0.5 M CaCl₂

Flowrate : 0.05 ml/min

Time (min)	PV	Type of elution fluid		
		DI H ₂ O	0.1 M CaCl ₂	0.5 M CaCl ₂
		HEDP Conc. (ppm)		
10.00	3.33	7163.75	5361.80	4850.27
20.00	6.67	1467.46	1533.82	1580.44
30.00	10.00	2019.74	880.83	538.03
40.00	13.33	2382.66	110.59	277.70
50.00	16.67	939.24	505.74	164.41
60.00	20.00	636.94	159.52	111.96
70.00	23.33	382.47	143.09	63.04
80.00	26.67	258.11	93.27	34.97
90.00	30.00	128.08	69.92	26.17
100.00	33.33	59.86	64.74	21.88
150.00	50.00	10.82	-	12.28
200.00	66.67	4.64	42.34	-
250.00	83.33	2.85	19.98	9.51
300.00	100.00	5.00	13.92	-
350.00	116.67	-	12.13	10.16
400.00	133.33	-	7.85	7.70
450.00	150.00	-	7.14	6.81
460.00	153.33	2.38	-	-
500.00	166.67	-	6.42	5.14
510.00	170.00	3.09	-	-
550.00	183.33	-	4.64	7.03
600.00	200.00	4.52	4.64	6.13
650.00	216.67	1.31	3.57	5.27
700.00	233.33	-	4.28	-
750.00	250.00	-	4.64	-
800.00	266.67	-	5.35	-
850.00	283.33	-	5.00	-

iii) Elution curves of 1:1 Ca-ATMP precipitates

Elution fluids : DI water, 0.1 M CaCl₂, 0.5 M CaCl₂

Flowrate : 0.05 ml/min

Time (min)	PV	Type of elution fluid		
		DI water	0.1 M CaCl ₂	0.5 M CaCl ₂
		ATMP Conc. (ppm)		
10	3.33	8466.04	3083.59	3156.02
20	6.67	2602.94	2840.99	1402.38
30	10.00	2570.34	1593.85	1335.11
40	13.33	2483.92	1588.68	1390.58
50	16.67	2398.16	1545.31	1362.83
60	20.00	2104.86	1496.13	1284.13
70	23.33	1811.56	1416.69	1256.52
80	26.67	1511.66	1342.12	1158.84
90	30.00	1367.25	1311.06	1090.61
100	33.33	1141.32	1206.70	1067.30
150	50.00	490.55	868.69	841.34
200	66.67	243.68	462.74	466.93
250	83.33	169.10	209.19	235.94
300	100.00	58.43	84.47	105.39
350	116.67	23.39	38.03	67.53
400	133.33	4.50	21.59	34.16
450	150.00	3.08	12.76	15.16
500	166.67	-	10.18	3.73
550	183.33	2.04	7.47	3.40
600	200.00	4.06	4.17	2.75
650	216.67	1.00	2.97	-
700	233.33	1.00		-
800	266.67	-	2.75	2.64
950	316.67	-	2.35	2.75
1000	333.33	-	2.35	3.08

CURRICULUM VITAE

Name : Ms. Piyarat Wattana

Birth Date : May 14, 1974

Nationality : Thai

University Education:

1991-1995 Bachelor's Degree of Science in Chemical Engineering
Chemical Technology, Faculty of Science,
Chulalongkorn University, Bangkok, Thailand.