# STUDY ON THE BENZOXAZINE MONOMERS AND THEIR APPLICATION FOR ION EXTRACTION MATERIAL

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## บทคัดย่อ

สุทธินันท์ พงษ์ธรรมรักษ์ : การศึกษาเบนซอกซาซีนมอนอเมอร์และการประยุกต์สำหรับ การเป็นวัสคุจำแนกไอออน (Study on the Benzoxazine Monomers and Their Application for Ion Extraction Material) อ.ที่ปรึกษา : ศ. ฮัทสุโอะ อิชิคะ (Prof. Hatsuo Ishida) และ คร. สุวบุญ จิรชาญชัย 42 หน้า ISBN 974-638-444-9

อนุพันธ์ของเบนซอกซาซีนสามประเภท ได้แก่ 3-เมธิล-3,4-ไดไฮโดร-6,8-ไดเมธิล-2เอช-1.3-เบนซอกซาซีน (3-methyl-3,4-dihydro-6,8-dimethyl-2H-1,3-benzoxazine, 3-เมธิล-3,4-ไดไฮโดร-6-เมธิล-8-โบรโม-2เอช-1,3-เบนซอกซาซีน (3-methyl-3,4dihydro-6-methyl-8-bromo-2H-1,3-benzoxazine, Bm2) และ 3-เมธิล-3,4-ไคไฮโคร-6-เมธิล-2เอช-1,3-เบนซอกซาซีน (3-methyl-3,4-dihydro-6-methyl-2H-1,3-benzoxa zine, Bm3) ซึ่งมีหมู่ฟังก์ชั่นต่างกันที่ตำแหน่งออโท (ortho) ของวงเบนซีน ได้ถูกสังเคราะห์ขึ้น และวิเคราะห์ด้วยโปรตอนนิวเคลียร์แมกเนติกเรโซแนนซ์ (<sup>1</sup>H-NMR) คาร์บอนนิวเคลียร์แมก เนติกเรโซแนนซ์ (13C-NMR) และฟเรียร์ทรานฟอร์มอินฟราเรคสเปกโตรสโคปี (FT-IR) การ ศึกษาสมบัติของสารประกอบหลัก-รอง (Host-guest compound) โดยการจำแนกใอออน พบ ว่า Bm1 ซึ่งเป็นเบนซอกซาซีนที่มีหมู่เมธิลที่ตำแหน่งออโท มีการจับไอออนสูงกว่า Bm3 ซึ่งไม่มี หมู่ฟังก์ชั่นใด ๆ ที่ตำแหน่งนั้น อันเป็นผลกระทบของสเตอริก (Steric effect) โดยที่โมเลกุลเบน ซอกซาซีนประเภท Bml มีการจัดเรียงกลุ่มของโมเลกุล (Molecular assembly) แบบหลวม (flexible) ในขณะที่โมเลกลของเบนซอกซาซีนประเภท Bm3 มีการจัดเรียงแบบแน่น (rigid) จึงทำให้เกิดความแตกต่างในการรับไอออน

#### **ABSTRACT**

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KEYWORDS: Benzoxazine monomer/ Ion extraction / Inclusion properties /

Structurally controlled benzoxazine derivatives

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Structurally controlled benzoxazine derivatives, i.e., 3-methyl-3,4-dihydro-6,8-dimethyl-2H-1,3-benzoxazine (Bm1), 3-methyl-3,4-dihydro-6-methyl-8-bromo-2H-1,3-benzoxazine (Bm2) and 3-methyl-3,4-dihydro-6-methyl-2H-1,3-benzoxazine (Bm3) with different functional group at ortho position on the phenol ring were successfully prepared as determined from <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, FTIR spectra as well as elemental analysis. Bm1, which has the bulky group of methyl at ortho position, gives higher ion extraction percentage than Bm3, which has no functional group at ortho position. In the case of Bm1, the steric effect of benzoxazine structure leads to the loosely packed molecular assembly which induces the higher ion affinity.

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