

## REFERENCES

- Admanson, A.W. Physical Chemistry of Surface. 5th ed., John Wiley & Sons, Inc., 1990.
- Baxter, J.E., Davidson, S., Hageman, H.J., McLauchlan, K.A. and Stevens, D.G. The Photo-induced Cleavage of Acylphosphine Oxides. J. CHEM. SOC. CHEM. COMMUN. (1987): 73-75.
- Brewmeyer Jr., W.F. Text Book of Polymer Science. 3rd ed., John Wiley & Sons, Inc., 1984.
- Cooper, A.R. Determination of Molecular Weight. John Wiley & Sons, Inc., 1990.
- Decker, C., Elzaoul, B. and Decker, D. Kinetic Study of Ultrafast Photopolymerization Reactions. Journal of Macromolecular Science Pure and Applied Chemistry Part A33, No.1 (1996): 173-190.
- Feldman, D. and Babalata, D. Synthetic Polymers Technology, Properties Applications. 1st ed., Chapman and Hall, 1996.
- Flory, P.J. Principles of Polymer Chemistry. Cornell University press, 1953.
- Herman, Uno, T., Kubono, A., Umemoto, S., Kikutani, T. and Okui, N. Effect of Molecular Weight and Chain End Groups on Crystal Forms of Poly(vinylidene fluoride) Oligomers. Polymer 38(7) (1997): 1677-1683.
- Iida, K., Nohara, T. and Totani, Kazuyuki. Molecular Orientation and Photocurrent of Alkyl-aromatic Polyimide Films Prepared by Vapor Deposition Polymerization. Japanese Journal of Applied Physics 28(12) (1989) : 2552-2555.

- Iijima, M. and Takahashi, Y. Techniques for Vapor Deposition Polymerization Preparation of Organic Functional Thin Films. Oyo Buturi 66(10) (1997): 1084-1088.
- Iijima, M., Takahashi, Y. and Oishi, Y. Preparation of Ultrathin Films of Aromatic Polyimides by Vapor Deposition Polymerization from N-Silylated Aromatic Diamines or Aromatic Tetracarboxylic Dithioanhydride. Journal of Polymer Science Part A: Polymer Chemistry 29 (1991): 1717-1723.
- Ito, Y., Hikira, M. and Kimura, T. Effect of Degree of Imidization on Polyimide Thin Films Prepared by Vapor Deposition Polymerization on the Electrical Conduction. Japanese Journal of Applied Physics 29(11) (1992) : 1129-1131.
- Jou, Jwo-Hue., Cheng, Chen-Li., Jou, Chin-Yi, Emily. and Yang, Chang-Mou, Arnold., Characterization of Vapor Deposition Polymerized Polyimide Thin Films. Journal of Polymer Science : Part B : Polymer Physics 34 (1996) : 2239-2246.
- Kissin., Y.V. Molecular Weight Distribution of Linear Polymers: Detailed Analysis from GPC Data. Journal of Polymer Science Part C 27(1969): 207-235.
- Orihashi, Y., Iwata, R., Taniguchi, I. and Itaya, A. Vapor Deposition Photopolymerization of Long-chain Vinyl Compounds - Polymerizability and Molecular - Orientation of N-octadecyl acrylate. Chemistry of Materials 7(2) (1995): 324-332.
- Pola, J. and Morit, H. UV Laser-induced Gas-phase Polymerization of Trimethyl(propynyloxy)silane Tetrahedron Letters. 38(44) (1997): 7809-7812.
- Roffey, C.G. Photopolymerization of Surface Coating. John Wiley & Sons, Inc., 1982.

- Rose, A. and Rose, E. The Condensed Chemical Directory. 6th ed., Reinhold New York, 1961.
- Sperling, L.H. Introduction to Physical Polymer Science. 2nd ed., John Wiley & Sons, Inc., 1993.
- Takahashi, Y., Matsuzaki, K., Iijima, M., Fukada, E., Tsukahara, S., Murakami, Y. and Maesono, A. Determination of Evaporation Rate and Vapor-pressure of Organic Monomers used for Vapor-deposition Polymerization Japanese Journal of Applied Physics Part 2-Letters. 32(6B) (1993): L875-L878.
- Takahashi, Y., Iijima, M., Oishi, Y., Kakimoto, M. and Imai, Y. Preparation of Ultrathin Films of Aromatic Polyamides and Aromatic Poly(amide-imides) by Vapor Deposition Polymerization. Macromolecules 24 (1991): 3543-3546.
- Takahashi, Y., Iijima, M., Fukada and E. Pyroelectricity in Poled Films of Aromatic Polyurea Prepared by Vapor Deposition Polymerization. Japanese Journal of Applied Physics 28(12) (1989): 2245-2247.
- Tamada , M., Omichi, H. and Okui, N. Change of Molecular Orientation with Post-polymerization of a Thin Film of *N*-methylolacrylamide Prepared with VDP Thin solid films 274 (1996) : 66-69.
- Tamada , M., Omichi, H. and Okui, N. Preparation of Polyvinylcarbazole Thin Film with Vapor Deposition Polymerization Thin solid films 268(1995) : 18-21.
- Tamada , M., Omichi, H. and Okui, N. Real-time In-situ Observation of Vapor Deposition of *N*-methylolacrylamide with IR-RAS. Thin solid films 260 (1995) : 168-173.
- Tamada , M., Omichi, H. and Okui, N. Vapor Deposition Polymerization of *N*-methylolacrylamide. Thin solid films 251 (1994) : 36-39.

Tamada, M., Asano, M., Yoshida, M. and Kumakura, M. Formation of a Thin Film of Poly(octadecyl methylmethacrylate) using the Physical Vapor Deposition Technique. Polymer 32(11) (1991) : 2064-2069.

## APPENDICES

### APPENDIX I PHYSICAL CHARACTERIZATION

#### I.1 Deposition Weight

**Table I.1.1** Effect of Substrate Temperature

Substrate Temperature (°C)	Deposition Weight (mg)
-5.0	17.8
0.0	15.6
3.0	4.5
5.0	3.4
10.0	2.5

**Table I.1.2** Effect of Amount of Photoinitiator

Amount of Photoinitiator (mmole)	Deposition Weight (mg)
0.430	10.0
0.861	14.0
1.722	13.2
5.167	17.2
8.612	16.5

**Table I.1.3** Effect of Deposition Time

Deposition Time (hours)	Deposition Weight (mg)
4.0	3.4
8.0	17.4
12.0	15.4
18.0	16.4 ( $\pm$ 1.48)
24.0	20.1

**Table I.1.4** Effect of Flow Rate

Flow Rate of gas mixture (ml N <sub>2</sub> /min)	Deposition Weight (mg)
49.0	1.2
82.0	3.5
123.0	15.6
179.0	16.1
243.5	16.8
331.0	17.5

**I.2    The Estimated Thickness**

$$L = K \times \text{deposition weight (mg)} \quad (\mu\text{m})$$

$$K = \frac{10}{\rho \text{ (g.cm}^{-3}\text{)} A \text{ (cm}^2\text{)}} \quad (\text{mg} \cdot \mu\text{m})$$

$$\rho = 1.0865 - (6.19 \times 10^{-4} \times T) + (1.36 \times 10^{-7} \times T^2) \quad (\text{g.cm}^{-3})$$

whereas;   L   = estimated thickness

K   = normal factor

$\rho$    = temperature dependent density of polystyrene

A   = surface area

T   = experimental temperature (C)

when,      T   = 25 C and A      = 11.35 cm<sup>2</sup>

then,      K   = 0.822  $\mu\text{m} \cdot \text{mg}^{-1}$

**Table I.2.1** Effect of Substrate Temperature

Substrate Temperature (°C)	Deposition Weight (mg)	Estimated Thickness (μm)
-5.0	17.8	14.647
0.0	15.6	12.837
3.0	4.5	3.703
5.0	3.4	2.798
10.0	2.5	2.057

**Table I.2.2** Effect of Amount of Photoinitiator

Amount of Photoinitiator (mmole)	Deposition Weight (mg)	Estimated Thickness (μm)
0.430	10.0	8.311
0.861	14.0	11.520
1.722	13.2	10.862
5.167	17.2	14.153
8.612	16.5	13.577

**Table I.2.3** Effect on Deposition Time

Deposition Time (hours)	Deposition Weight (mg)	Estimated Thickness ( $\mu\text{m}$ )
4.0	3.4	2.798
8.0	17.4	14.318
12.0	15.4	12.672
18.0	16.4 ( $\pm 1.48$ )	13.510 ( $\pm 1.22$ )
24.0	20.1	16.540

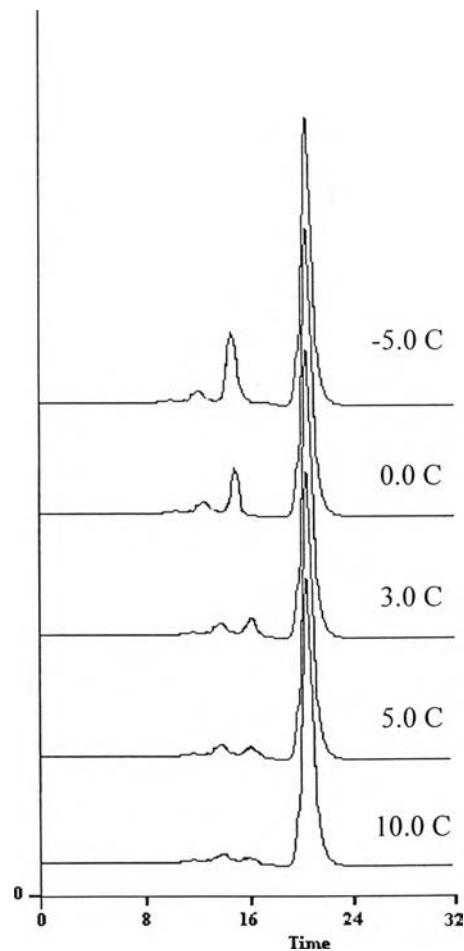
**Table I.2.4** Effect on Flow Rate

Flow Rate of gas mixture (ml N <sub>2</sub> /min)	Deposition Weight (mg)	Estimated Thickness ( $\mu\text{m}$ )
49.0	1.2	0.987
82.0	3.5	2.88
123.0	15.6	12.837
179.0	16.1	13.248
243.5	16.8	13.824
331.0	17.5	14.400

## APPENDIX II

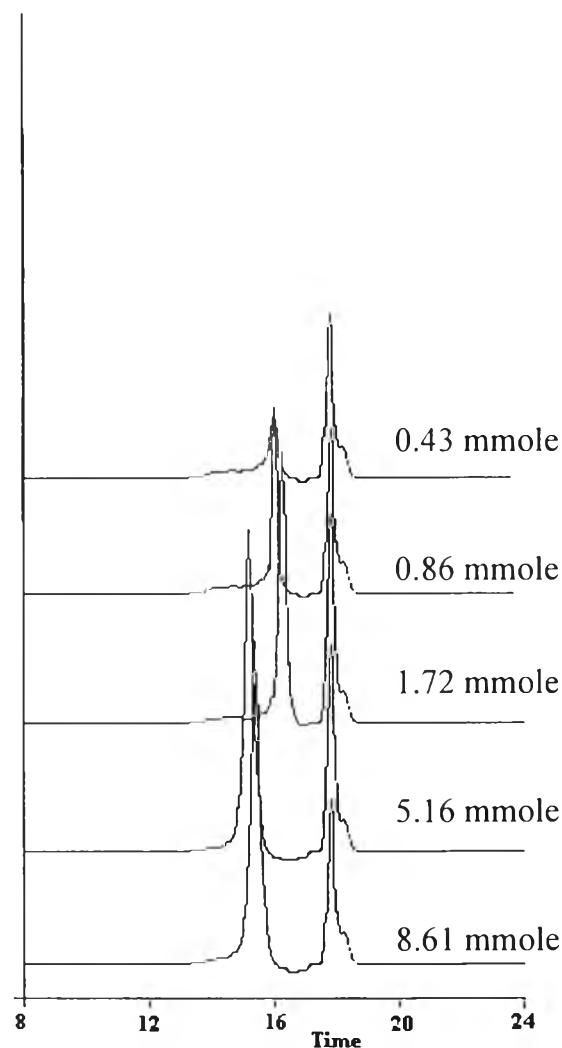
### GEL PERMEATION CHARACTERIZATION

#### II.1 Effect of Substrate Temperature



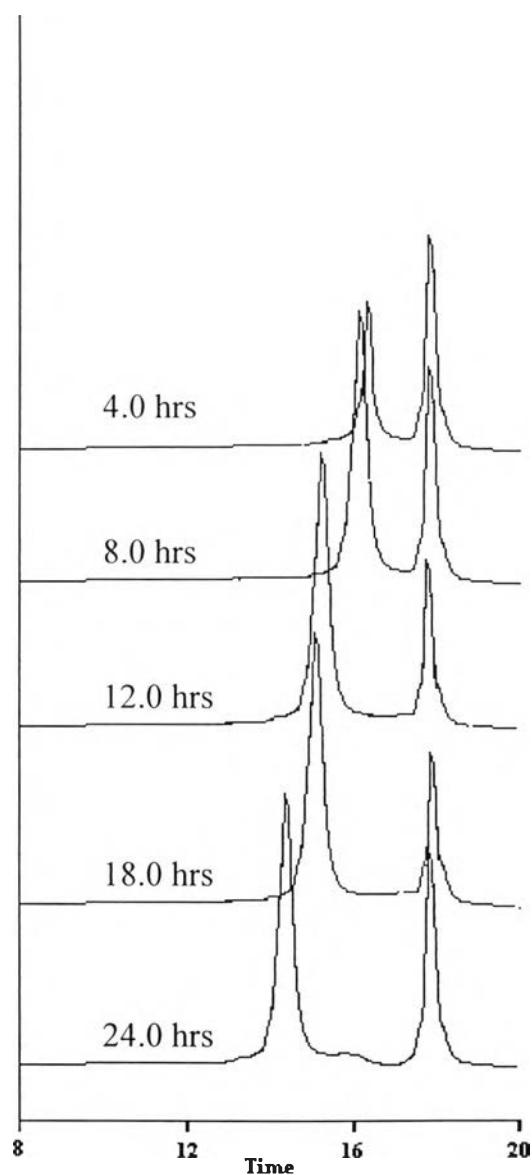
**Figure II.1** Gel permeation chromatogram of deposited film at the substrate temperature of -5.0 - 10.0 °C

## II.2 Effect of Amount of Photoinitiator



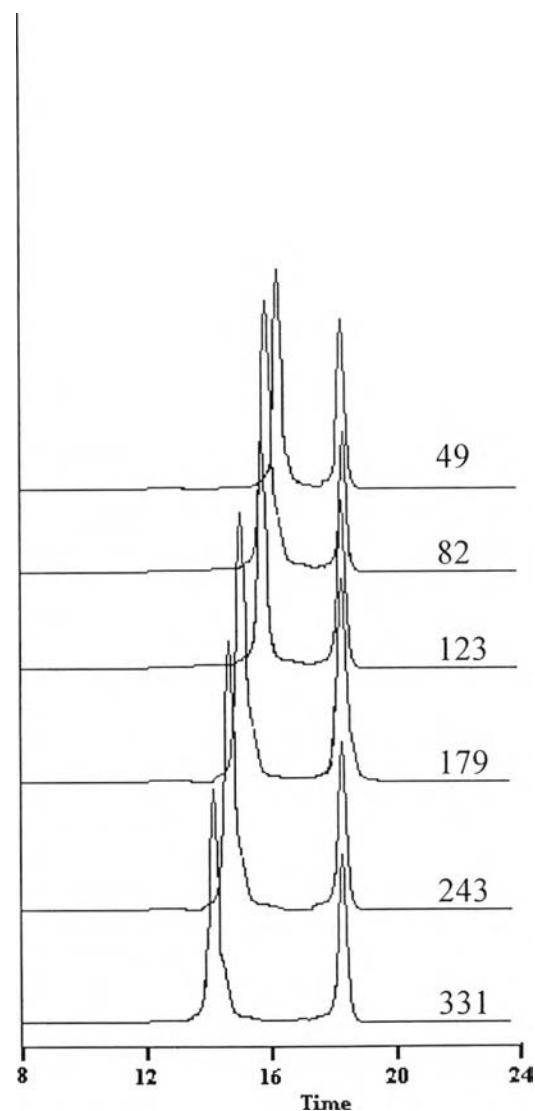
**Figure II.2** Gel permeation chromatogram of deposited film with varying amount of photoinitiator.

### II.3 Effect of deposition Time



**Figure II.3** Gel permeation chromatogram of deposited film at deposition time of 4.0 - 24.0 hours.

#### II.4 Effect of Flow Rate



**Figure II.4** Gel permeation chromatogram of deposited film at  $\text{N}_2$  flow rate of 49 - 331  $\text{ml.N}_2/\text{min.}$

## CURRICULAR VITAE

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