EFFECT OF EXTERNAL SILANE DONORS ON ZIEGLER-NATTA CATALYST FOR POLYMERIZATION OF PROPYLENE

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ABSTRACT

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The effect of external silane donors of propylene polymerization, with a MgCl₂-supported TiCl₄ catalyst activated by Al(C₂H₅)₃, has been investigated. Five types of alkoxy silanes were used as external donors. The degree of isotactic index and activity depended on the types of alkoxy silane and amount of external donor used. Iso and cyclic structures of external donors enhanced the isotactic index. Addition of small amounts of external donors led to a significant increase in activity and isotactic index. At constant external donor content, an increase of Al/Ti resulted in an increase in activity and a decrease in isotactic index. Moreover, the activity of the polymer increased while the isotactic index decreased with increasing hydrogen concentration. The number average molecular weight (\overline{M}_n) of the polypropylene decreased as a result of hydrogen addition for the polymerization in the absence and presence of all electron donors.

บทคัดย่อ

นิพัทธ์ ถ้าเลิศลักษณชัย : ศึกษาผลของเอ็กทานอล ใชเลนโดเนอร์ต่อตัวเร่งปฏิกิริยาซิก เลอร์แนตด้าของโพลีโพร ใพลีน (Effect of Extenal Silane Donors on Ziegler-Natta Catalyst for Polymerization of Propylene) อ.ที่ปรึกษา : ศ. สมชาย โอสุวรรณ คร. รัตนวรรณ มกรพันธุ์ และ คร. สุรชา อุดมศักดิ์ 72 หน้า, ISBN 974-334-141-2

การวิจัยนี้ใด้ศึกษาผลของเอ็กทานอลใชเลนโดเนอร์ (External silane donors) ของ การทำโพลิเมอไรซ์โพรไพลีนที่ใช้ตัวเร่งปฏิกิริยาไททาเนียมเตตระคลอไรด์บนแมกนีเซียมคลอ ไรด์ชับพอร์ทซึ่งถูกทำให้ว่องไวโดยไตรเอททิลอลูมินัม โดยใช้อัลคอกซี่ใชเลน 5 ชนิดเป็นเอ็กทานอลโดเนอร์ ผลที่มีต่อความว่องไวของปฏิกิริยาและดัชนีไอโซ่แทคทิค (isotactic index) จะ มากหรือน้อยขึ้นอยู่กับชนิดของอัลคอกซี่ใชเลนและปริมาณเอ็กทานอลโดเนอร์ที่ใช้เดิมลงไปโครงสร้างไอโซและไซคลิกของเอ็กทานอลโดเนอร์ช่วยส่งเสริมค่าดัชนีไอโซ่แทคทิค การเติมสาร เอ็กทานอลโดเนอร์เพียงเล็กน้อยจะทำให้ความว่องไวของปฏิกิริยาและค่าดัชนีไอโซ่แทคทิคเพิ่ม ขึ้น อัตราส่วนของอลูมินัมต่อไทเทเนียมเพิ่มขึ้นที่ความเข้มข้นของเอ็กทานอลโดเนอร์คงที่จะทำให้ความว่องไวของปฏิกิริยาเพิ่มขึ้นและดัชนีไอโซ่แทคทิคลดลง ยิ่งไปกว่านั้นด้วยการเพิ่มความเข้มข้นไฮโดรเจนทำให้ความว่องไวของปฏิกิริยาเพิ่มขึ้นขณะที่ดัชนีไอโซ่แทคทิคลดลง การเติมโฮโดรเจนจะทำให้น้ำหนักโมเลกุลเฉลี่ยโดยจำนวน (\overline{M}_s) ลดลงอีกด้วยทั้งที่มีการเติมเอ็กทานอลโดเนอร์เดะไม่มีการเติมเอ็กทานอลโดเนอร์

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