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APPENDICES

Appendix A Silatrane Synthesis

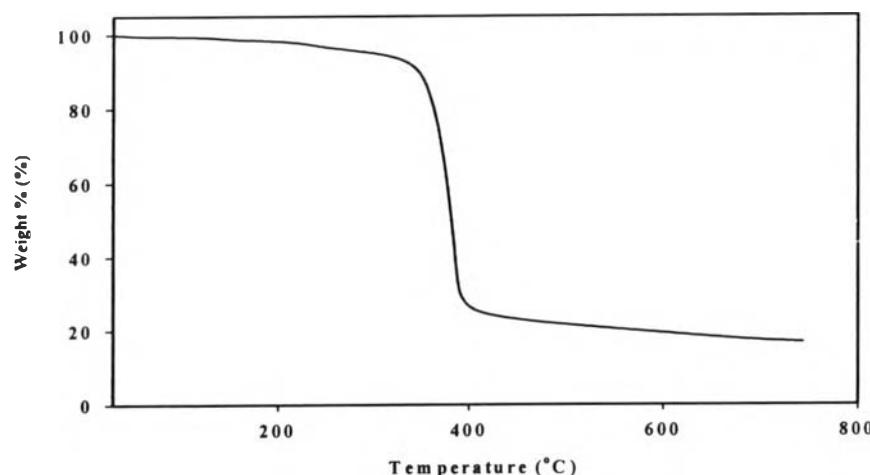
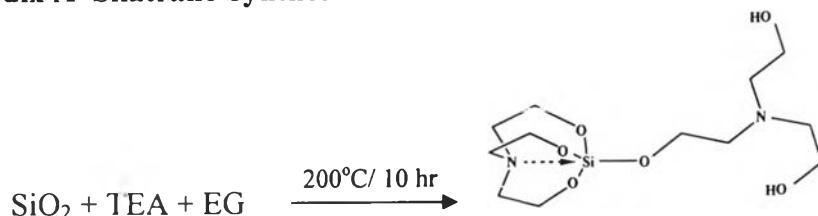


Figure A1 TGA of silatrane

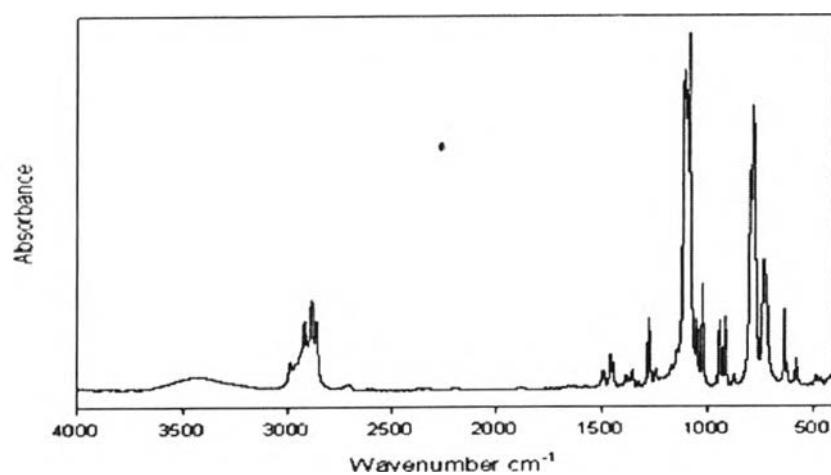


Figure A2 FTIR of silatrane

Table A1 FTIR spectrum of silatrane

Peak Positions (cm-1) Assignments		Peak Positions (cm-1)	Assignments
3100-3700	b, ν O-H	1276	m, ν C-O
2800-3000	s, ν C-H	1040-1180	b & vs, ν Si-O
2750-2670	w, Si<--N)	786	vs, δ Si-O-C
1445, 1459, 1493	m, δ C-H	735	s, δ Si-O-C
1351	w, ν C-N	576	w, Si<--N

FAB⁺-MS showed the highest m/e at 669 of $\text{Si}_3((\text{OCH}_2\text{CH}_2)_3\text{N})_4\text{H}^+$ and 100% intensity at 323 of $\text{Si}((\text{OCH}_2\text{CH}_2)_3\text{N})_2\text{H}^+$.

Appendix B Titanium Glycolate synthesis

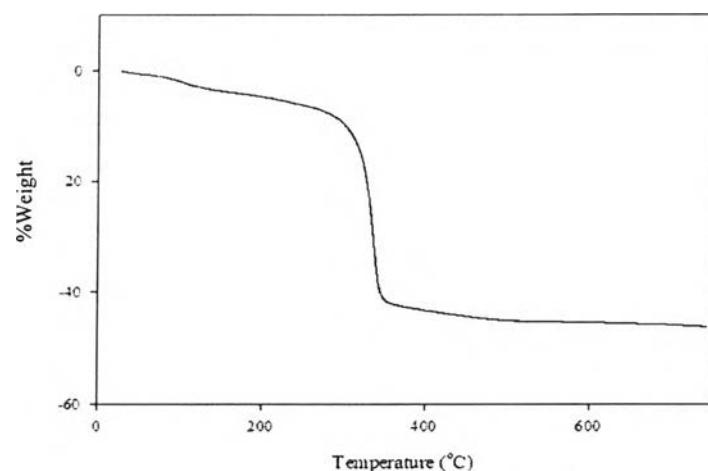
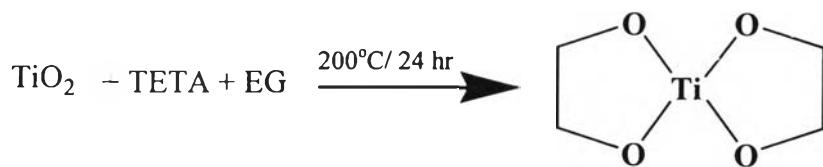


Figure B1 TGA of titanium glycolate

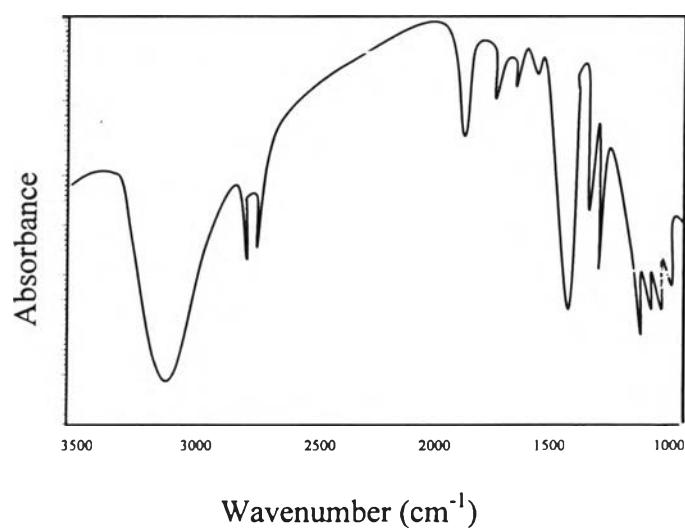


Figure B2 FTIR of silatrane

Table B1 FTIR spectrum of titanium glycolate

Peak Positions (cm-1)	Assignments	Peak Positions (cm-1)	Assignments
3100-3700	b, ν O-H	1073	s, νC-Ö
2860-2986	s, νC-H	1049	s, νSi-O
1244-1275	m, νC-N	1021	s, νC-O
1170-1117	bs, νSi-O	785, 729	s, νSi-O-C
1093	s, νSi-O-C	579	w, Si<---N

FAB⁺-MS: showed m/e 169 with 8.5% intensity of Ti(OCH₂CH₂O)₂.

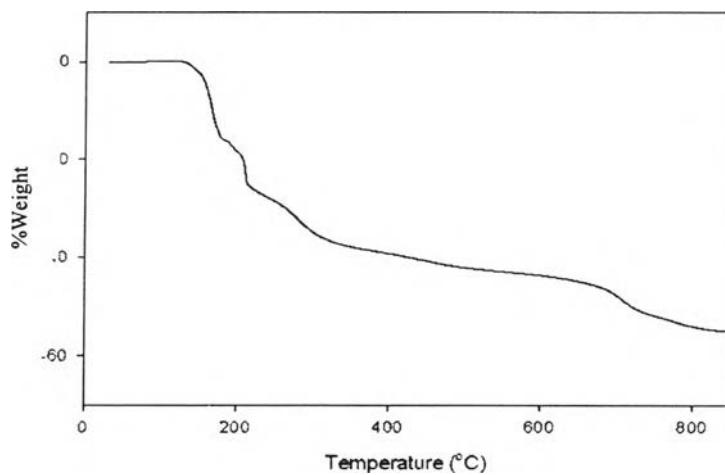
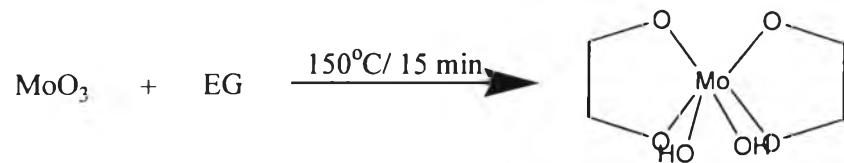
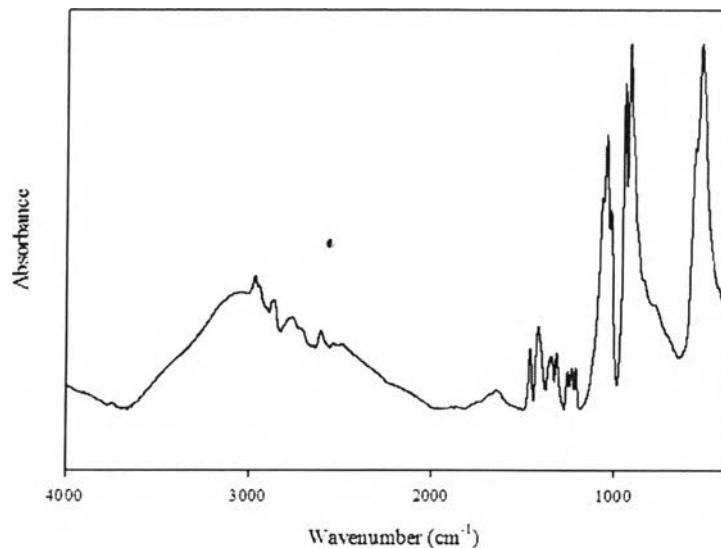
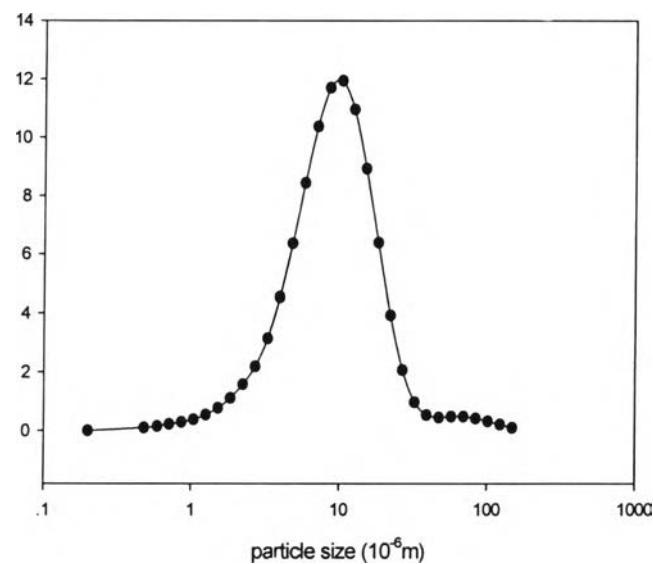
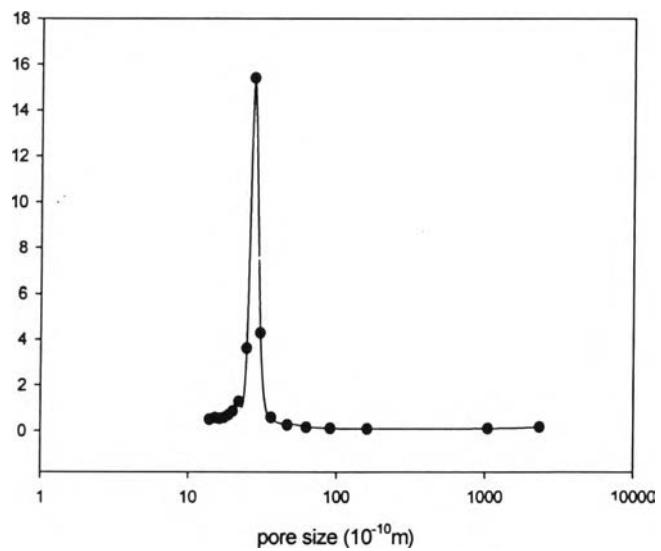
Appendix C Molybdenum Glycolate Synthesis**Figure C1** TGA of molybdenum glycolate**Figure C2** FTIR of silatrane

Table C1 FTIR spectrum of silatrane

Peak Positions (cm-1)	Assignments	Peak Positions (cm-1)	Assignments
3500-3000	s,vOH	946	vMo-O-C
2968-2860	vC-H	533	vM-O
1344-1258	vCH-OH		

Appendix D Micro and Macro Structure of MCM-41**Figure D1** Particle size distribution of MCM-41**Figure D2** Pore size distribution of MCM-41

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2. Thanabodeekij, N., Sathupunya, M., Jamieson, A.M., and Wongkasemjit, S. (2003) Correlation of sol-gel processing parameters with microstructure and properties of a ceramic product. Material Characterization, 50, 325-337.
3. Thanabodeekij, N., Tanglumlert, W., Gulari, E. and Wongkasemjit, S. (2005) Synthesis of Ti-MCM-41 Directly from Silatrane and Titanium Glycolate and Its Activity. Applied Organometallic Chemistry, 19, 1047-1054.
4. Thanabodeekij, N., Sadthayanon, S., Gulari, E. and Wongkasemjit, S. (2005) Extremely High Surface Area of Ordered Mesoporous MCM-41 by Atrane Route. Materials Chemistry and Physics, in press.
5. Thanabodeekij, N., Gulari, E. and Wongkasemjit, S. (2005) $\text{Bi}_{12}\text{TiO}_{20}$ Synthesized Directly from Bismuth (III) Nitrate Pentahydrate and Titanium Glycolate and Its Activity. Powder Technology, in press.

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Proceedings:

1. Thanabodeekij, N., Tanglumlert, W., Gulari, E. and Wongkasemjit, S. (2004, December 6-7) Effect involving the formation of extremely high surface area MCM-41. Proceeding of Micro-and Mesoporous Mineral Phases 2004, Rome, Italy.

Presentation:

1. Thanabodeekij, N., Tanglumlert, W., Gulari, E. and Wongkasemjit, S. (2004, December 6-7) Effect involving the formation of extremely high surface area MCM-41. Proceeding of Micro-and Mesoporous Mineral Phases 2004, Rome, Italy.